



The superconductivity at 18 K in LiFeAs system

X.C. Wang, Q.Q. Liu, Y.X. Lv, W.B. Gao, L.X. Yang, R.C. Yu, F.Y. Li, C.Q. Jin *

Institute of Physics, Chinese Academy of Science, Beijing 100080, China

ARTICLE INFO

Article history:

Received 1 September 2008

Received in revised form

18 September 2008

Accepted 27 September 2008 by X.C. Shen

Available online 14 October 2008

PACS:

74.10.+v

81.05.Bx

62.50.+p

Keywords:

A. Superconductors

B. Chemical synthesis

C. Crystal structure & symmetry

ABSTRACT

The recent discovery of superconductivity in iron arsenide compounds $RFeAsO$ (R = rare earth) or AFe_2As_2 (A = alkaline earth) has attracted great attention due to the unexpected high T_c in the system containing ferromagnetic elements like Fe. Similar to high T_c cuprates, the superconductivity in iron arsenide is related to a layered structure. Searching for new superconductors with [FeAs] layer, but of simpler structure will be of scientific significance either to build up new multilayered superconductors that may reach higher T_c or to study the mysterious underlined superconducting mechanism in iron arsenide compounds. Here we report that a new superconducting iron arsenide system LiFeAs was found. The compound crystallizes into a structure containing [FeAs] conducting layer that is interlaced with Li charge reservoir. Superconductivity was observed with T_c up to 18 K in the compounds.

© 2008 Elsevier Ltd. All rights reserved.

1. Introduction

Since the discovery of iron pnictide superconductor [1] with superconducting transition temperature (T_c) 26 K for $LaFeOAs$ [2] a series of rare earth iron quaternary pnictide oxides $RFeAsO$ superconductors (R = Ce, Pr, Nd, Sm, ...) were found [3–7]. The superconducting transition temperature in $RFeAsO$ system (or termed “1111” according to the composition ration) was quickly raised above 50 K [6]. More recently a new quaternary oxygen free iron arsenide $(Ba, K)Fe_2As_2$ [8] superconductors with T_c 38 K was found that can be alternatively termed as “122” system. Like the [CuO2] plane that plays a key role in high T_c superconducting copper oxide [9,10], the [FeAs] layer [11] is believed crucial to support superconductivity in the iron arsenide superconductors. But different from high T_c cuprates that belong to the category of strongly correlated charge transfer type Mott compounds, the layered iron arsenide is an itinerant metal/semi-metal. The unexpected high transition temperature in this itinerant system challenges the conventional BCS mechanism based on electron phonon coupling scenario. Searching for new iron arsenide superconductors with simple structure will be of general interest either to unveil the underlined superconducting mechanism [11–15] or to further enhance T_c . Here we report that a new superconducting iron arsenide system LiFeAs (termed “111”)

was found. LiFeAs crystallizes into a Cu_2Sb type tetragonal structure [16] containing [FeAs] layer with an average iron valence Fe^{+2} like those for “1111” or “122” parent compounds. Superconductivity with T_c up to 18 K was found in the compounds.

2. Experimental

The LiFeAs compounds have been synthesized with the assistance of high pressure sintering. The starting materials of Li (99.9%) plus FeAs are mixed according to the nominal formula $Li_{1-x}FeAs$ or $Li_{1+x}FeAs$ ($0 \leq x \leq 0.6$). The FeAs precursors are synthesized from high purity Fe and As powders that are sealed into an evacuated quartz tube. The mixtures are sintered at 800 °C for 10 h for several times to assure single phase nature. Since the compositions of LiFeAs are either hygroscopic or easy to react with oxygen or nitrogen, the processes were performed in a glove box protected with high purity Ar. We synthesized the “111” type LiFeAs with the assistance of high pressure high temperatures since high pressure can effectively prevent lithium from oxidizing or evaporation upon heating in addition to accelerate the reaction. The pellets of mixed starting materials wrapped with gold foil were sintered at 1–1.8 GPa, 800 °C for 60 min followed by quenching from high temperature before releasing pressure. Samples were characterized by x-ray powder diffraction with a Mac Science diffractometer. Diffraction data were collected with 0.02° step and 15 s/step. Rietveld analysis has been performed using the program GSAS software package. The magnetic properties of the samples were measured using a superconducting quantum interference

* Corresponding author. Tel.: +86 10 82649163; fax: +86 10 82649531.
E-mail address: jin@aphy.iphy.ac.cn (C.Q. Jin).

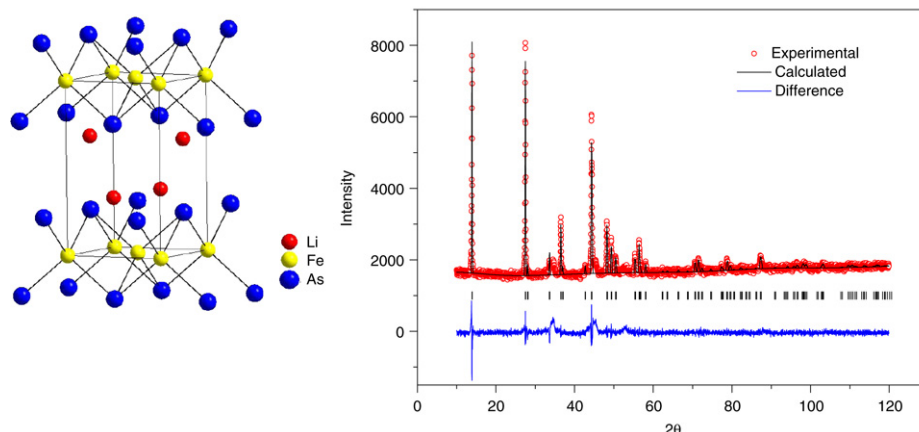


Fig. 1. The powder x ray diffraction pattern as well as Rietveld refinements for a LiFeAs sample crystallizing into a structure that is featured by alternative [FeAs] layer being interlaced with Li similar to the infinite layer CaCuO_2 .

device (SQUID) (quantum design). The electric conductivity was measured using the standard four probe method.

3. Results & discussions

Fig. 1 shows the x-ray diffraction patterns of LiFeAs sample that can be well indexed into a nearly pure Cu_2Sb type tetragonal structure [16] of layered [FeAs] nature with space group $P4/nmm$. As shown in Fig. 1 the [FeAs] conducting layer is interlaced with Li charge reservoir in LiFeAs. Using Reitveld method the x ray diffraction patterns were also refined based on the structure model [16] as shown in Fig. 1 & Table 1. This structure is similar to Fe_2As but with the interlayered Fe being replaced with Li. The lattice parameters obtained for LiFeAs are $a = 3.77 \text{ \AA}$, $c = 6.36 \text{ \AA}$. Compared with “1111” RFeAsO or “122” AFe_2As_2 , both c axis or the ab plane is considerably shrunk for “111” LiFeAs. Fig. 2 shows the temperature dependence of the electric conductivity of samples with nominal composition $\text{Li}_{1-x}\text{FeAs}$. The parent LiFeAs shows Pauli metallic behavior with a pronounced curvature at wide temperature range below room temperature, indicating the electron correlation behavior. It could be the intrinsic properties of LiFeAs or it may result from the composition inhomogeneous that leads to a little bit Li instoichiometry in LiFeAs compounds. It is worth of mentioning that no evidence has been observed so far for spin density wave (SDW)-related phase transition in $\text{Li}_{1\pm x}\text{FeAs}$ system that usually results in resistance drop at the transition temperature as observed for “1111” type RFeAsO or AFe_2As_2 ($A = \text{Ba}, \text{Sr}$). The electric conductivity was significantly enhanced in samples with nominal composition $\text{Li}_{1-x}\text{FeAs}$ indicating the improved carrier doping. The superconducting transition was clearly observed for samples of nominal composition $\text{Li}_{1-x}\text{FeAs}$ ($x = 0, 0.2, 0.4$) with T_c ranging from 16 K to 18K as shown in Fig. 2. The almost constant superconducting transition temperature implies the point superconducting phase feature of $\text{Li}_{1\pm x}\text{FeAs}$ that is in contrast with high T_c cuprates where T_c usually systematically evolves with carrier density. Fig. 3 shows the dc magnetic susceptibility of sample with nominal composition $\text{Li}_{0.6}\text{FeAs}$ measured in both zero field cooling (ZFC) & field cooling (FC) mode with a $H = 10 \text{ Oe}$. The large Meissner signal ($> 10\%$) indicates the bulk superconducting nature of the sample.

Here we address several interesting features for this “111” type LiFeAs superconducting system compared with “1111” type RFeAsO or “122” type AFe_2As_2 . There is no evidence so far, either from the electric conductivity or magnetic susceptibility measurements, showing the SDW for LiFeAs superconducting system that is quite different from “1111” RFeAsO or “122” AFe_2As_2

Table 1

Crystallographic data for LiFeAs.

Atom	Site	x	y	z	Occupancy
Li	2c	0	0.5	0.343	1
Fe	2a	0	0	0	1
As	2c	0	0.5	0.735	1

Then Rietveld refinement of x ray diffraction patterns for LiFeAs at room temperature. Space group: $P4/nmm$. Unit-cell dimensions: $a = b = 3.7715(2) \text{ \AA}$, $c = 6.3574(3) \text{ \AA}$; $R_{wp} = 8.79\%$, $R_p = 5.62\%$, where R_{wp} & R_p are the R factors for the weighted profile, profile, respectively. The refinement range of 2θ was 10° – 120° . $\text{CuK}\alpha$ radiation was used. The values in parentheses represent the standard deviation in the last digit. The isotropic thermal parameters were set to be constant for all atoms. The occupancy at each site was fixed as 1.

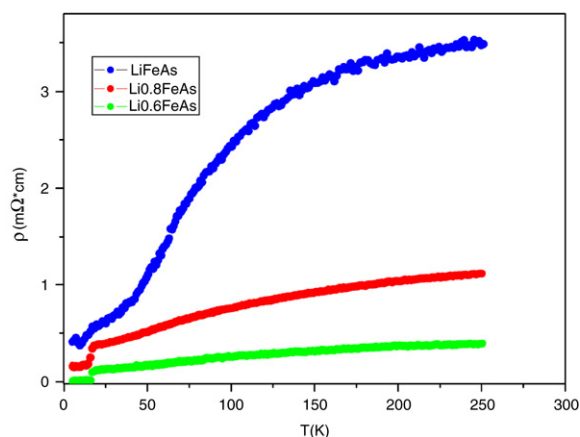


Fig. 2. The temperature dependence of resistivity for LiFeAs compounds showing superconducting transition up to 18 K for the sample of nominal composition $\text{Li}_{0.6}\text{FeAs}$ with the metallic behavior at normal state.

where the superconductivity appears once the SDW transition was suppressed. It is true that the absence of the anomaly can not completely rule out the presence of SDW order in the parent compound: one possibility is the absence of SDW related structure phase transition observed in “1111” systems [11] that makes the effects on resistivity much smaller. However, if the absence of SDW order in present “111” LiFeAs system is confirmed by further experiments, it will raise an important question: SDW may not be a prerequisite for introducing superconductivity in this [FeAs] compound. Then what is the “glue” leading to superconducting pairing in LiFeAs? The space group of LiFeAs is $P4/nmm$, i.e. the same as “1111” type RFeAsO , so there is no in plane shift for the

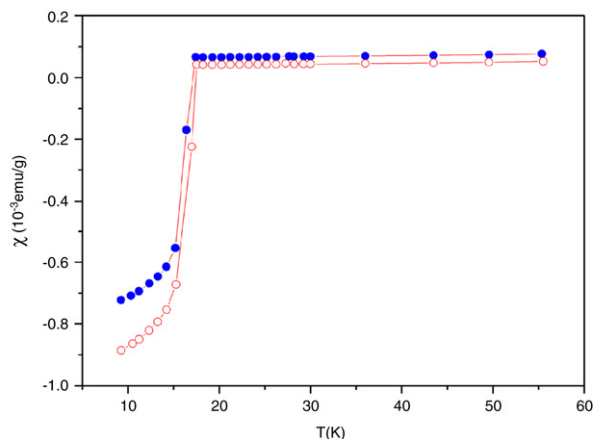


Fig. 3. The dc susceptibility of sample with nominal composition $\text{Li}_{0.6}\text{FeAs}$ in both ZFC & FC mode, indicating the bulk superconducting nature.

neighboring interlayer components (charge reservoir layer). On the other hand, the alkaline earth layer shifts half periodicity in the ab plane for the body-centered “122” type AFe_2As_2 with space group $I4/mmm$. This feature makes LiFeAs more like an “infinite layer structure” comparing with the prototype ACuO_2 for high T_c cuprate [17,18]. Consequently the “111” type LiFeAs will be ideal for the theoretical studies due to its plain crystal structure form; or it can be core component to build up multilayered iron arsenide superconductors that may lead to higher T_c as the case for high T_c superconductor cuprates.

Notes: The primary results of this work about the superconducting LiFeAs system have been uploaded as [arXiv:0806.4688](https://arxiv.org/abs/0806.4688).

Two weeks later two other groups also reported the superconductivity in LiFeAs ([arXiv:0807.2228](https://arxiv.org/abs/0807.2228); [arXiv:0807.2274](https://arxiv.org/abs/0807.2274)).

Acknowledgments

The NSF and Ministry of Science & Technology of China are thanked for financial support (grant number: 2005CB724402; 2006CB601002; 2007CB925003). We are grateful to Profs. L.Yu, Zhongxian Zhao, & J. S. Zhou for helpful discussions.

References

- [1] Y. Kamihara, H. Hiramoto, M. Hirano, R. Kawamura, H. Yanagi, T. Kamiya, H. Hosono, *J. Am. Chem. Soc.* 128 (2006) 10012.
- [2] Y. Kamihara, T. Watanabe, M. Hirano, H. Hosono, *J. Am. Chem. Soc.* 130 (2008) 3296.
- [3] H. Takahashi, K. Igawa, K. Arii, Y. Kamihara, M. Hirano, H. Hosono, *Nature* 453 (2008) 376.
- [4] X.H. Chen, T. Wu, G. Wu, R.H. Liu, H. Chen, D.F. Fang, *Nature* 453 (2008) 761.
- [5] H.H. Wen, G. Mu, L. Fang, H. Yang, X.Y. Zhu, *Euro. Phys. Lett.* 82 (2008) 17009.
- [6] Z.-A. Ren, J. Yang, W. Lu, W. Yi, X.-L. Shen, Z.-G. Li, G.-C. Che, X.-L. Dong, L.-L. Sun, F. Zhou, Z.-X. Zhao, *Europhys. Lett.* 82 (2008) 57002.
- [7] G.F. Chen, Z. Li, D. Wu, G. Li, W.Z. Hu, J. Dong, P. Zheng, J.L. Luo, N.L. Wang, *Phys. Rev. Lett.* 100 (2008) 247002.
- [8] M. Rotter, M. Tegel, D. Johrendt, *Phys. Rev. Lett.* 101 (2008) 107006.
- [9] J.G. Bednorz, K.A. Müller, *Z. Phys. B* 64 (1986) 189.
- [10] M. Karppinen, H. Yamauchi, *Mater. Sci. Eng. R* 26 (1999) 5.
- [11] Clarina de la Cruz, Q. Huang, J.W. Lynn, Jiying Li, W. Ratcliff, J.L. Zarestky, H.A. Mook, G.F. Chen, J.L. Luo, N.L. Wang, Pengcheng Dai, *Nature* 453 (2008) 899.
- [12] X. Dai, Z. Fang, Y. Zhou, F.C. Zhang, [arXiv:0803.3982](https://arxiv.org/abs/0803.3982) (2008).
- [13] Paul.M. Grant, *Nature* 453 (2008) 1000.
- [14] Qiang Han, Yan Chen, Z.D. Wang, [arXiv:0803.4346](https://arxiv.org/abs/0803.4346) (2008).
- [15] Z.P. Yin, S. Lebedev, M.J. Han, B. Neal, S.Y. Savrasov, W.E. Pickett, [arXiv:0804.3355](https://arxiv.org/abs/0804.3355) (2008).
- [16] Von.R. Juza, K. Langer, *Z. Anorg. Allg. Chem.* 356 (1968) 253.
- [17] T. Siegrist, S.M. Zahurak, D.W. Murphy, R.S. Roth, *Nature* 334 (1988) 231.
- [18] M.G. Smith, A. Manthiram, J. Zhou, J.B. Goodenough, J.T. Markert, *Nature* 351 (1991) 549.